

SUPPORTING INFORMATION

Probing Electrostatic Interactions along the Reaction Pathway of a Glycoside Hydrolase: Histidine Characterization by NMR Spectroscopy

Mario Schubert, David K. Y. Poon, Jacqueline Wicki, Chris A. Tarling, Emily M. Kwan, Jens E. Nielsen, Stephen G. Withers, and Lawrence P. McIntosh

Supplemental Table S1: Chemical shifts and $^1J_{\text{NH}}$ couplings for the histidine residues in CexCD (30 °C, pH 6.5).

Residue	Protein Form	$^{13}\text{C}^{\delta 2}\text{H}$ (ppm)	$^{13}\text{C}^{\epsilon 1}\text{H}$ (ppm)	$^{15}\text{N}^{\delta 1}\text{H}$ (ppm)	$^{15}\text{N}^{\epsilon 2}\text{H}$ (ppm)	$^1J_{\text{N}\delta 1\text{H}}$ (Hz) ^b	$^1J_{\text{N}\epsilon 2\text{H}}$ (Hz) ^b
His80	apo	- /7.12	137.9/7.42	202.5 ^a / -	182.4 ^a / -	-	-
	2FCb	129.3/6.83	138.6/6.64	172.1/13.80	246.9/ -	98±1	-
	XbIso	- /6.63	137.5/6.71	172.4/13.85	243.6/ -	- ^c	-
	XbIm	- /6.74	140.1/7.04	169.9/13.32	244.1/ -	97±2	-
His85	apo	118.4/6.91	139.2/8.52	247.6/-	167.7/11.83	-	-
	2FCb	118.4/6.88	139.0/8.43	247.5/ -	167.5/11.80	-	101±2 ^d
	XbIso	- /6.85	138.5/8.35	- / -	167.0/11.75	-	-
	XbIm	- /6.79	138.9/8.50	248.1/ -	166.6/11.71	-	97±4
His107	apo	118.6/6.28	135.8/8.27	184.8/16.74	170.7/13.52	89±2	96±2
	2FCb	118.8/6.36	136.0/8.30	184.6/16.68	170.6/13.50	92±3	101±2
	XbIso	- /6.39	135.9/8.27	183.2/16.46	170.2/13.47	94±2	99±2
	XbIm	- /6.37	135.9/8.26	183.0/16.42	170.2/13.52	90±2	99±2
His114	apo	118.8/3.79	135.7/8.28	181.7 ^a / -	177.4 ^a / -	-	-
	2FCb	118.8/3.76	135.7/8.23	181.4 ^a / -	177.4 ^a / -	-	-
	XbIso	- /3.76	135.6/8.30	181.0 ^a / -	177.1 ^a / -		
	XbIm	- /3.76	135.6/8.30	181.0 ^a / -	177.1 ^a / -		
His205	apo	120.0/7.59	138.2/7.70	182.8 ^a / -	171.2 ^a / -	-	-
	2FCb	119.5/7.23	136.4/7.75	188.0/17.64	164.3/10.22	85±5	102±2
	XbIso	- /7.69	138.9/7.74	182.4/16.80	173.7/13.79	- ^c	- ^c
	XbIm	- /7.32	137.8/7.51	179.6/16.00	179.6/14.29	97±2	98±4

^aSpecific assignments of the $^{15}\text{N}^{\delta 1}$ and $^{15}\text{N}^{\epsilon 2}$ chemical shifts for protonated imidazoles could not be obtained from a multiple bond ^{15}N -HSQC spectrum (Fig. 3), and thus the downfield chemical shift was arbitrarily tabulated as that from the $^{15}\text{N}^{\delta 1}$ nucleus.

^bExcept where indicated, $^1J_{\text{NH}}$ couplings were measured with IPAP spectra.

^cNot determined due to overlap or weak signals.

^dMeasured from a ^{15}N -HSQC spectrum without ^{15}N decoupling.

Supplemental Table S2: Predicted pKa values for CexCD and 2FCb-CexCD.

His	Form ^a	Exp. pKa	Calc. PropKa ^b pKa	Calc. WHAT IF ^c pKa	Decomposition ^{d,e}		
					Desolvation	H-bond or Background	Ionizable
H80	apo	7.9	5.2		-5.0	1.6 (D123)	-1.2 (K47) 1.5 (D123) 1.8 (E233)
					9.8	-2.3	0.9
	2FCb	< 2.8		6.7	-3.0	1.2	2.2
H85	apo	< 4	0.90		-3.3	0.04 (E127)	-2.4 (R136)
					4.4	-1.3	-0.5
	2FCb	< 2.8		4.3	-1.3	-0.1	-0.6
H107	apo	>10.4	4.9		-4.9	1.60 (E52)	1.7 (E52)
					8.4	-3.4	2.7
	2FCb	>10.1		8.9	-3.4	3.2	2.8
H114	apo	8.1	7.0		-0.3	0.8 (D64)	0.0
					6.5	-1.0	0.6
	2FCb	8.1		6.6	-1.0	0.6	0.7
H205	apo	>10.4	8.8		-5.4	2.1 (E233) 1.6 (D235)	1.7 (E233) 2.4 (D235)
					11.3	-2.0	-0.1
	2FCb	>10.1		< 0	-2.5	-2.7	< 0

^aUsing the pdb files for apo-CexCD (2EXO.pdb) and 2FCb-CexCD (1EXP.pdb)

^bCalculated using the PropKa webserver (<http://propka.chem.uiowa.edu/>) (1)

^cpKa calculations were carried out with the WHAT IF pKa routines as described previously (2). The structural co-ordinate pdb files for apo-CexCD (2EXO.pdb) and 2FCb-CexCD (1EXP.pdb) were regularized using WHAT IF (3), and non-protein atoms, except those of the bound inhibitor, were removed. Charge and radii parameters for protein atoms were taken from the OPLS force field (4). Ligand topology files and charges on ligand atoms were calculated using the PRODRG web-server (5). Radii were assigned manually based on the OPLS force field (4). The WHAT IF pKa calculation routines use an algorithm developed by Hoofstede *et al.* (6) to find the optimal hydrogen-bond network for every protonation state needed in the pKa calculation scheme (7). The electrostatic interaction energies for the calculation then were obtained by solving the linear form of the Poisson-Boltzmann equation using the DelphiII program (8) with the following parameters: temperature 298.15K, ionic strength 0.144 M, ion exclusion radius 2.0 Å, solvent dielectric constant 80, protein dielectric constant.

^ePropKa pKa values calculated with an intrinsic pKa 6.5 of histidine, corrected for desolvation, for sidechain and backbone hydrogen bonds, and for interactions with ionizable groups in presumed charged states.

^dWHAT IF pKa values calculated with an intrinsic pKa 6.3 of histidine, corrected for desolvation and for interactions with background (non-titratable) and ionizable groups (at pH = pKa).

Supporting References

1. Li, H., Robertson, A. D., and Jensen, J. H. (2005) Very fast empirical prediction and interpretation of protein pKa values, *Proteins* 61, 704-721.
2. Nielsen, J. E., and Vriend, G. (2001) Optimizing the hydrogen-bond network in Poisson-Boltzmann equation-based pK(a) calculations, *Proteins* 43, 403-412.
3. Vriend, G. (1990) WHAT IF: a molecular modeling and drug design program, *J. Mol. Graphics* 8, 52-56.
4. Jorgensen, W. L., and Tirado-Rives, J. (1988) The OPLS potential functions for proteins: energy minimizations for crystals for cyclic peptides and crambin, *J. Am. Chem. Soc.* 110, 1657-1666.
5. Schuttelkopf, A. W., and van Aalten, D. M. (2004) PRODRG: a tool for high-throughput crystallography of protein-ligand complexes, *Acta Crystallogr. D* 60, 1355-1363.
6. Hooft, R. W., Sander, C., and Vriend, G. (1996) Positioning hydrogen atoms by optimizing hydrogen-bond networks in protein structures, *Proteins* 26, 363-376.
7. Yang, A. S., Gunner, M. R., Sampogna, R., Sharp, K., and Honig, B. (1993) On the calculation of pKa's in proteins, *Proteins* 15, 252-265.
8. Nicholls, A., and Honig, B. (1991) A rapid finite difference algorithm, utilizing successive over-relaxation to solve the Poisson-Boltzmann equation, *J. Comp. Chem.* 12, 435-445.